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Use the following passage and table to answer questions 11-14. A student studying electron configurations in atoms created Table 1 in order to summarize the various facts she learned about principal energy levels, energy sublevels, and orbitals. 11. Based on Table 1 what is the relationship between the principal energy level number ( $n$ )

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1. C 4. B 7. C. 2. C 5. C 8. A. 3. D 6. C 9.  
B. 10. Answers may vary. 11.  
Observations C and G are chemical

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