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___ 5. Mendeleev is credited with developing the first successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. ___ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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Solutions Manual Chemistry: Matter and Change • Chapter 5 73 CHAPTER 5 SOLUTIONS MANUAL Chapter 5 Assessment pages 166–169 Section 5.1 Mastering Concepts 34. Define the following terms. a. frequency Frequency is the number of waves that pass a given point per second. b. wavelength Wavelength is the shortest distance between

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5. Write the noble-gas notation for iodine. 6. Identify each atom. a. 1s22s22p1 b. [Ar]4s1 carbon sulfur 7. Write electron-dot structures for the following atoms. Chapter Assessment a. neon b. hydrogen 28 Chemistry: Matter and Change Chapter 5 c. d.

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Chapter 5 - Electrons in Atoms - 5 Assessment - Page 156: 106 Answer The atomic mass of chlorine is very far from a whole because a weighted average of atomic masses of all of its isotopes is computed in determining its atomic mass.

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